

UUCMS No.

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B.M.S. COLLEGE FOR WOMEN AUTONOMOUS
BENGALURU-560004

SEMESTER END EXAMINATION-APRIL/MAY- 2023

M.Sc. in Chemistry-I Semester

ORGANIC CHEMISTRY-I

Course code: MCH102T

Time: 3 hrs

QP Code: 11008

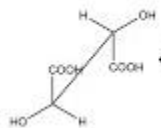
Max.Marks:70

Instruction: Answer Question No.1 and any FIVE of the remaining.

1. Answer any TEN questions

(2×10 =20)

- Illustrate Tautomerism with example?
- What is cross conjugation? Give an example.
- Discuss non-classical carbocation with an example.
- State Curtin-Hammett principle.
- Convert the following projection into Newman and Fischer projections.



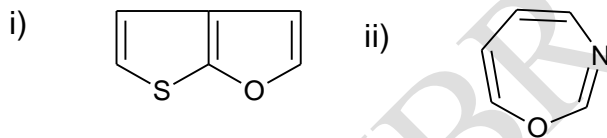
- Illustrate Prelog's rule.
- Write the mathematical expression for Taft equation and elaborate the terms.
- Outline any two methods of generation of carbanion.
- Give any two example of an ambidentate nucleophile.
- Draw the structural formulae of the following carbohydrates.
 - Lactose
 - Gentiobiose
- Give any one synthesis of coumarin.
- What are deoxy sugars? Give an example.

2. a) Give an account of generation and reactivity of nitrenes.
b) Illustrate hyperconjugation with suitable example.
c) Write a note on cyclodextrins. **(4+3+3=10)**

3. a) Outline the generation, structure and stability of free radicals.
b) Give an account of Hammett equation and linear free energy relationship. **(5+5=10)**

4. a) State and explain Cram's rule with suitable example.
b) Write the synthesis and reaction of imidazole.
c) Write a note on conformational analysis of cyclobutane and cyclopentane. **(4+3+3=10)**

5. a) Elucidate the structure of sucrose.
b) Write the IUPAC nomenclature of following heterocyclic compounds:



(5+5=10)

6. a) Outline briefly the bonding patterns in C-60 fullerenes.
b) Explain the method of determination of configuration of aldotetrose
c) List out the three major differences between S_N^1 and S_N^2 mechanisms.

(4+3+3=10)

7. a) Write a note on:
i) Anchimeric effect ii) Steric effect.
b) Write the differences between enantiotopic and diastereotopic groups and faces with proper examples.

c) Outline the synthesis of aldonic acid. **(4+3+3=10)**

8. a) Describe briefly the synthesis of Uronic acid.
b) Write a short note on ionophore and micelles.
c) Discuss SE_2 mechanism with suitable examples.

(4+3+3=10)